

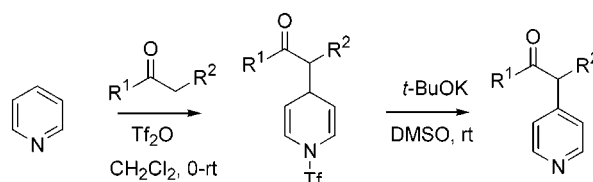
Regiospecific Synthesis of  
4-(2-Oxoalkyl)pyridinesAlan R. Katritzky,\* Suoming Zhang,<sup>†</sup> Thomas Kurz,<sup>§</sup> and Mingyi WangDepartment of Chemistry, Center for Heterocyclic Compounds, University of Florida,  
Gainesville, Florida 32611-7200Peter J. Steel<sup>‡</sup>

Department of Chemistry, University of Canterbury, Christchurch, New Zealand

katritzky@chem.ufl.edu

Received June 1, 2001

## ABSTRACT

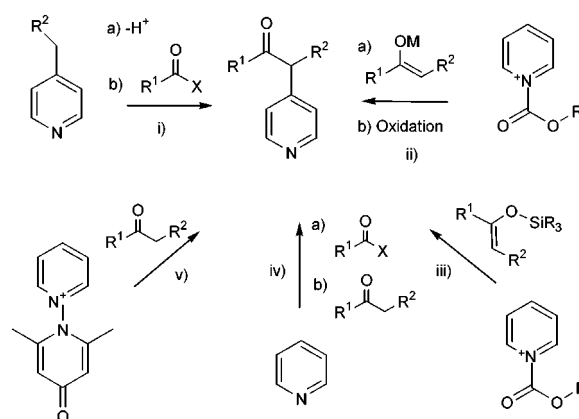


A new and operationally simple method has been developed for the regiospecific syntheses of 4-(2-oxoalkyl)pyridines from ketones and pyridine in good yields, using triflic anhydride to activate the pyridine ring.

Functionalization of nitrogen heterocycles constitutes a powerful tool for the synthesis of natural products and bioactive substances. The regioselective introduction of a  $\beta$ -oxo-alkyl group into the 4-position of a pyridine has attracted considerable attention (Scheme 1): (i) classically, the preparation of 4-(2-oxoalkyl)pyridines involves the treatment of a deprotonated 4-alkylpyridine with an ester or acid chloride.<sup>1</sup> (ii) Comins and Brown have prepared 4-(2-oxoalkyl)pyridines by the treatment of 1-oxycarbonylpyridinium salts with lithium or titanium enolates followed by reaction with sulfur and 5% Pd/C in naphthalene.<sup>2</sup> (iii) Akiba and co-workers reacted trimethylsilyl enol ethers with 1-ethoxycarbonylpyridinium chloride to afford 1-ethoxycarbonyl-4-(2-oxoalkyl)-1,4-dihydropyridines regioselectively.<sup>3</sup> (iv) Doering and McEwen treated ketones with pyridine and

benzoyl chloride, or acetic anhydride, to obtain 4-substituted dihydropyridines.<sup>4</sup> (v) Our group developed regiospecific syntheses of 4-substituted (including 4-( $\beta$ -oxoalkyl)) pyridines via *N*-(2,6-dimethyl-4-oxo-1,4-dihydropyrid-1-yl)pyridinium salt.<sup>5</sup> We now present a practical route for the regiospecific synthesis of 4-(2-oxoalkyl)pyridines using triflic

## Scheme 1



<sup>†</sup> Present address: Neugen Corporation, 35 Northeast Industrial Road, Branford, CT 06405.

<sup>§</sup> Present address: Institute of Pharmacy, University of Hamburg, Bundesstrasse 45, D-10146 Hamburg, Germany.

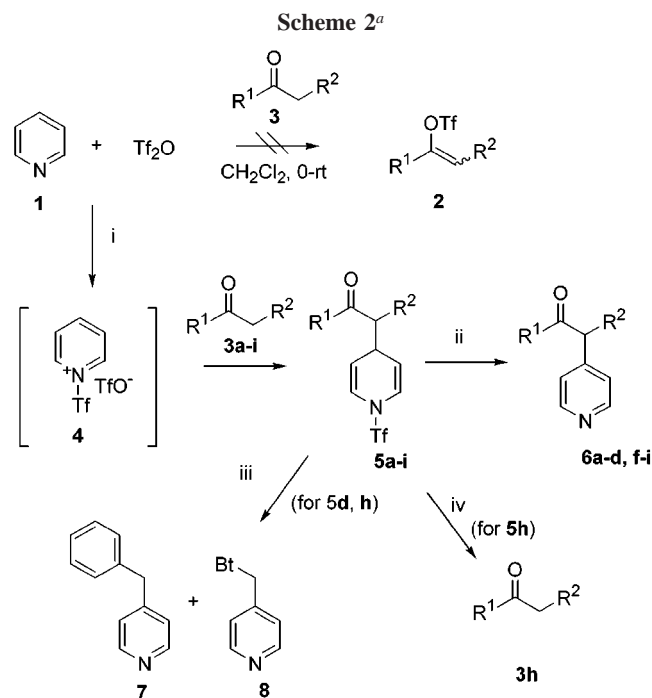
<sup>‡</sup> Email: p.steel@chem.canterbury.ac.nz.

(1) (a) Katritzky, A. R.; Rees, C. W. *Comprehensive Heterocyclic Chemistry*; Pergamon Press: Oxford, 1984; Vol. 2. (b) Katritzky, A. R.; Rees, C. W.; Scriven, E. F. V. *Comprehensive Heterocyclic Chemistry II*; Pergamon Press: Oxford, 1996; Vol. 2.

(2) Comins, D. L.; Brown, J. D. *Tetrahedron Lett.* **1984**, 25, 3297.

anhydride to activate the pyridine ring, which complements the previous routes and offers an advantage in terms of convenience and yield.

**4-(2-Oxoalkyl)-1,4-dihydropyridines 5a–i.**<sup>6</sup> Treatment of ketones **3a–i** with pyridine **1** (4 equiv) and triflic anhydride (3 equiv) in CH<sub>2</sub>Cl<sub>2</sub> at 0 °C gave 4-(2-oxoalkyl)-1,4-dihydropyridines **5a–i** (in an average yield of 88%) rather than enol triflates **2** (Scheme 2). No regioisomeric 2-(2-



<sup>a</sup> Conditions and reagents: (i) CH<sub>2</sub>Cl<sub>2</sub>, 0 °C; (ii) *t*-BuOK, DMSO or *t*-BuOH, rt; (iii) TBAF in THF, 70 °C; (iv) KOH in DMSO, 70 °C.

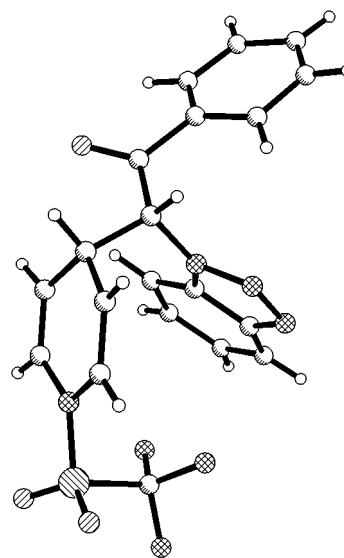
oxoalkyl)-1,2-dihydropyridines were detected from the NMR spectra of the crude products **5a–i**. In the case of **5i**, the starting material **3i** was recovered in 15% yield. The structures of these novel 1,4-dihydropyridines **5a–i** were supported by their NMR spectra, and structure **5f** was confirmed by X-ray crystallography (Figure 1). This crystal structure determination confirmed the regiochemistry of the

(3) (a) Akiba, Y.; Nishihara, Y.; Wada, M. *Tetrahedron Lett.* **1983**, *24*, 5269. (b) Wada, M.; Nishihara, Y.; Akiba, Y. *Tetrahedron Lett.* **1985**, *26*, 3267.

(4) Doering, W. von E.; McEwen, W. E. *J. Am. Chem. Soc.* **1951**, *73*, 2104.

(5) (a) Katritzky, A. R.; Beltrami, H.; Keay, J. G.; Rogers, D. N.; Sammes, M. P.; Leung, C. W. F.; Lee, C. M. *Angew. Chem., Int. Ed. Engl.* **1979**, *18*, 792. (b) Lee, C. M.; Sammes, M. P.; Katritzky, A. R. *J. Chem. Soc., Perkin Trans. 1* **1980**, 2458.

(6) **Typical Procedure for the Preparation of 5.** To a cooled (0 °C) solution of a ketone **3** (10 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (50 mL) was added pyridine (dried over NaOH, 3.2 mL, 40 mmol), and then triflic anhydride (5.0 mL, 30 mmol) was added dropwise over 30 min at 0 °C. The ice bath was removed, and the reaction mixture stirred at room temperature overnight. The reaction mixture was washed with saturated NH<sub>4</sub>Cl (2 × 50 mL) and H<sub>2</sub>O (2 × 50 mL). The organic layer was dried over MgSO<sub>4</sub>. The crude products were purified by recrystallization or chromatography to afford the desired 1,4-dihydropyridines **5** in good to excellent yields.



**Figure 1.** Perspective view of the X-ray crystal structure of **5f**.

reaction. The dihydropyridine ring has a conformation similar to those previously reported for three other 1-triflyl-1,4-dihydropyridines.<sup>7</sup>

**Table 1.** 4-Substituted 1,4-Dihydropyridines **5**

entry	R <sup>1</sup>	R <sup>2</sup>	mp (°C)	yield (%)
<b>5a</b>	(CH <sub>3</sub> ) <sub>2</sub> CH	C <sub>6</sub> H <sub>5</sub>	44–45	96
<b>5b</b>	CH <sub>3</sub>	4-CH <sub>3</sub> OC <sub>6</sub> H <sub>4</sub>	47–48	86
<b>5c</b>	4-BrC <sub>6</sub> H <sub>4</sub>	C <sub>6</sub> H <sub>5</sub>	104–105	88
<b>5d</b>	C <sub>6</sub> H <sub>5</sub>	C <sub>6</sub> H <sub>5</sub>	119–120	98
<b>5e</b>	C <sub>6</sub> H <sub>4</sub> CH <sub>2</sub> CH <sub>2</sub>			81
<b>5f</b>	C <sub>6</sub> H <sub>5</sub>	Bt	125–126	95
<b>5g</b>	2-thienyl	Bt	138	85
<b>5h</b>	2-furyl	Bt	114–115	91
<b>5i</b>	(CH <sub>3</sub> ) <sub>3</sub> C	Bt	125–126	72 (15) <sup>a</sup>

<sup>a</sup> Recovered starting material in parentheses.

**4-(2-Oxoalkyl)pyridines 6a–d,f–i.**<sup>8</sup> 1,4-Dihydropyridines **5a–d,f–i** on treatment with *t*-BuOK in DMSO or *t*-BuOH afforded **6a–d**,<sup>9,10</sup> **f–i** (average 92% yield). Treatment of **5h** with KOH in DMSO at 70 °C gave **3h** in quantitative yield. Pyridine derivative **6e** was not obtained as a result of

(7) (a) Toscano, R. A.; Hernandez-Galindo, M. del C.; Rosas, R.; Garcia-Mellado, O.; Portilla, F. del R.; Amabile-Cuevas, C.; Alvarez-Toledano, C. *Chem. Pharm. Bull.* **1997**, *45*, 957. (b) Toscano, R. A.; Rosas, R.; Hernandez-Galindo, M. del C.; Alvarez-Toledano, C.; Garcia-Mellado, O. *Transition Met. Chem.* **1998**, *23*, 113.

(8) **Typical Procedure for the Preparation of 6.** To a solution of compound **5** (5 mmol) in DMSO (20 mL) was added *t*-BuOK (1.68 g, 15 mmol) at 10 °C. The reaction mixture was stirred at room temperature for 10–30 min. The reaction was quenched with brine and extracted with EtOAc. The organic layer was washed with H<sub>2</sub>O and dried over MgSO<sub>4</sub>. Recrystallization from hexanes–ethyl acetate afforded product **6** in 81–97% yields.

(9) Anders, E.; Will, W.; Stankowiak, A. *Chem. Ber.* **1983**, *116*, 3192.

(10) Reynolds, S.; Levine, R. *J. Am. Chem. Soc.* **1960**, *82*, 472.

the instability of compound **5e**. The structures of the pyridine derivatives **6a–d,f–i** were characterized by their NMR and mass spectra. In the <sup>1</sup>H NMR spectra of **6a–d,f–i** two characteristic doublets ( $J = \text{ca. } 5.5 \text{ Hz, } 2\text{H}$ ) at ca. 8.50 and 7.20 ppm were assigned to the pyridine ring and a characteristic singlet at ca. 6.0 or 7.5 ppm was assigned to the methine group attached to the pyridine moiety.

However, several other attempts to eliminate the triflate group from 4-(2-oxoalkyl)-1,4-dihydropyridines **5** under various reaction conditions, including (i) reflux with isopropylamine, (ii) NaOCH<sub>3</sub> in methanol or toluene, and (iii) NaH in THF, failed. Use of TBAF in THF afforded the desired 4-(2-oxoalkyl)pyridine **6f** in 55% yield. In the case of **5d,h**, the application of TBAF failed to afford the corresponding products **6d,h** and gave instead **7**<sup>11</sup> and **8**, respectively.

In conclusion, we have developed a new and operationally simple method for the regiospecific synthesis of 4-(2-oxoalkyl)pyridines from ketones and pyridine, using triflic anhydride to activate the pyridine ring. This method comple-

(11) Chia, W.-L.; Shiao, M.-J. *Tetrahedron Lett.* **1991**, *32*, 2033.

**Table 2.** 4-(2-Oxoalkyl) Pyridines **6**

entry	R <sup>1</sup>	R <sup>2</sup>	mp (°C)	yield (%)
<b>6a</b>	(CH <sub>3</sub> ) <sub>2</sub> CH	C <sub>6</sub> H <sub>5</sub>	167–168	91
<b>6b</b>	CH <sub>3</sub>	4-CH <sub>3</sub> OC <sub>6</sub> H <sub>4</sub>	114–115	81
<b>6c</b>	4-BrC <sub>6</sub> H <sub>4</sub>	C <sub>6</sub> H <sub>5</sub>	235–236	97
<b>6d</b>	C <sub>6</sub> H <sub>5</sub>	C <sub>6</sub> H <sub>5</sub>	117–118	95
<b>6f</b>	C <sub>6</sub> H <sub>5</sub>	Bt	154–155	90 (55) <sup>a</sup>
<b>6g</b>	2-thienyl	Bt	foam	91
<b>6h</b>	2-furyl	Bt	211–212	94
<b>6i</b>	(CH <sub>3</sub> ) <sub>3</sub> C	Bt	153–154	95

<sup>a</sup> Compound **5f** was treated with TBAF in THF at 70 °C.

ments the existing methods as it is highly regioselective and high-yielding.

**Supporting Information Available:** Characterization data for compounds **5a–d,f–i**, **6a–d,f–i**, **7**, **8**, and X-ray data for **5f**. This material is available free of charge via the Internet at <http://pubs.acs.org>.

OL010116F